

Additional Table S1- S9. Additional Figure S1 and S6

Supplementary Table 1. Physical studies and analytical data of ligand, ML^1 & ML^1L^2

Compound	Colour	Molecular formula	Molecular weight	Elemental Analysis Found(cal.)%				Molar conductance $\Omega^{-1} \text{cm}^2 \text{Mol}^{-1}$
				C	N	O	M	
Ligand	Light Yellow	$C_{18}H_{16}N_2O_2$	292.12	73.95	9.58	10.95	-	-
CuL¹	Dark Green	$C_{40}H_{40}N_4O_9Cu$	783.21	61.26	7.14	18.36	8.10	12
NiL¹	Light brown	$C_{40}H_{46}N_4O_{12}Ni$	832.25	57.64	6.72	23.03	7.04	13
CoL¹	Orange	$C_{40}H_{46}N_4O_{12}Co$	833.76	57.32	6.72	23.03	8.60	18
ZnL¹	Yellowish Brown	$C_{40}H_{42}N_4O_{10}Zn$	802.22	59.74	6.97	19.89	9.97	15
Cu L¹ L²	Black	$C_{31}H_{30}N_3O_8Cu$	636.14	58.53	6.61	20.12	9.99	23
Ni L¹ L²	Brown	$C_{31}H_{36}N_3O_{11}Ni$	684.17	54.33	6.13	25.68	8.56	19
Co L¹ L²	Red brown	$C_{31}H_{36}N_3O_{11}Co$	685.17	54.31	6.13	25.67	7.07	11
Zn L¹ L²	Dirty white	$C_{31}H_{32}N_3O_9Zn$	654.14	56.76	6.41	21.95	8.13	15

Supplementary Table 2. FT-IR absorption data of L, ML^1 & ML^1L^2

Compound	Lattice Water Molecule (cm^{-1})	$\nu(\text{CH}=\text{N})$ (cm^{-1})	$\nu(-\text{OCH}_3)$ (cm^{-1})	$\nu(\text{CH}_2=\text{CH}_2)$ (cm^{-1})	8 Hydroxy quinoline		$\nu(\text{M}-\text{N})$ (cm^{-1})
					$\nu(\text{M}-\text{N})$ (cm^{-1})	$\nu(\text{M}-\text{O})$ (cm^{-1})	
Ligand	-	1581	2833	2946	-	-	-
CuL¹	-	1577	2828	2927	-	-	473
NiL¹	3300	1576	2828	2927	-	-	473
CoL¹	3404	1577	2829	2895	-	-	495
ZnL¹	3394	1575	2829	2900	-	-	472
Cu L¹ L²	3442	1580	2832	2944	495	519	483
Ni L¹ L²	3435	1574	2829	2950	497	493	472
Co L¹ L²	3381	1573	2828	2952	496	519	493
Zn L¹ L²	-	1573	2828	2949	491	533	472

Supplementary Table 3. The Absorption spectrum and magnetic moment data analysis of the L, ML^1 & ML^1L^2

Compounds	Absorption(nm)	Band assignment	Geometry	μ_{eff} (BM)
Ligand(L)	270 369	INCT INCT	---	---
[Cu(L)]	270 302 439	INCT INCT ${}^2T_{2g} \rightarrow {}^2E_g$	Square planar	1.95
[Co(L)]	264 382 450 510	INCT INCT ${}^4T_{1g}(F) \rightarrow {}^4T_{1g}(P)$ ${}^4T_{1g}(F) \rightarrow {}^4A_{2g}(F)$	Tetrahedral	4.41
[Ni(L)]	260 342 381 470	INCT INCT ${}^3A_{2g}(F) \rightarrow {}^3T_{1g}(P)$ ${}^3A_{2g}(F) \rightarrow {}^3T_{1g}(F)$	Square planar	3.81
[Zn(L)]	264 307 392	INCT INCT LMCT	Square planar	----
[Cu L¹ L²]	261 317 440	INCT INCT ${}^2B_{1g} \rightarrow {}^2A_{1g}$	Distorted square planar	1.81
[Ni L¹ L²]	263 393 457 511	INCT INCT ${}^4T_{1g}(F) \rightarrow {}^4T_{1g}(P)$ ${}^4T_{1g}(F) \rightarrow {}^4A_{2g}(F)$	Tetrahedral	4.83
[Co L¹ L²]	268 309 392 550	INCT INCT ${}^3A_{2g}(F) \rightarrow {}^3T_{1g}(P)$ ${}^3A_{2g}(F) \rightarrow {}^3T_{1g}(F)$	Square planar	3.05
[Zn L¹ L²]	261 312 482	INCT INCT LMCT	Square planar	----

Supplementary Table 4. Mass spectral data of the L, ML¹ & ML¹L²

Compound	Cal.Mass (m/z)	Obt. Mass(m/z)
Ligand	292.12	294.05
CuL¹	783.21	781.99
NiL¹	832.25	833.28
CoL¹	833.76	835.03
ZnL¹	802.22	802.84
Cu L¹ L²	636.14	636.14
Ni L¹ L²	684.17	678.48
Co L¹ L²	685.17	683.43

Zn L ¹ L ²	654.14	647.24
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Supplementary Table 5. Crystalline size of the compounds

Compound	2θ of the maximum intensity peak (Degrees)	θ (Degrees)	Cos θ	β	Crystalline Size in (Å) $t = 0.9\lambda/\beta\cos\theta$
Ligand	16.6283	8.31415	0.9894	0.001391	100.7488
CuL ¹	16.723	8.3615	0.9893	0.00281	49.87665
NiL ¹	16.723	8.3615	0.9893	0.001882	74.45885
CoL ¹	16.4863	8.24315	0.9896	0.001796	78.0251
ZnL ¹	11.8465	5.92325	0.9946	0.002163	64.44712
Cu L ¹ L ²	16.84626	8.42313	0.9892	0.002159	64.92777
Ni L ¹ L ²	16.94653	8.47326	0.9890	0.002938	47.72123
Co L ¹ L ²	16.8964	8.4482	0.9891	0.004728	29.65052
Zn L ¹ L ²	16.8964	8.4482	0.9890	0.003416	41.03534

Supplementary Table 6. Inhibition indices of Schiff base ligand, ML¹ and ML¹L² complexes against *E. coli*, *S. aureus*, and *Candida albicans*

Sample Marking	Sample Concentration (μg/ml)	Zone of inhibition in (mm)		Sample Concentration (μg/ml)	Zone of inhibition in (mm)
		<i>S. aureus</i>	<i>E. coli</i>		
Standard	Tetracyclin	13	12	Fluconazole	12
Ligand		NA	10		2
CuL ¹		5	8		5
NiL ¹		NA	7		4
CoL ¹		7	9		8
ZnL ¹	100	6	6	100	9
Cu L ¹ L ²		12	12		10
Ni L ¹ L ²		10	9		6
Co L ¹ L ²		8	9		11
Zn L ¹ L ²		7	5		8

Supplementary Table 7. Anti-Inflammatory activity of Schiff base ligand, CuL¹ L² and CuL¹ complexes

Compounds	Concentration (µg/ml)					IC ₅₀ (µg/ml)
	20	40	60	80	100	
% of inhibition at 560 nm						
Standard	56.28	60.14	67.49	72.78	78.69	0.592
Ligand	32.23	45.86	49.62	66.91	71.42	53.58
CuL ¹ L ²	42.53	56.77	69.79	71	79.51	28.50
CuL ¹	38.34	43.6	59.39	61.65	72.93	48.14

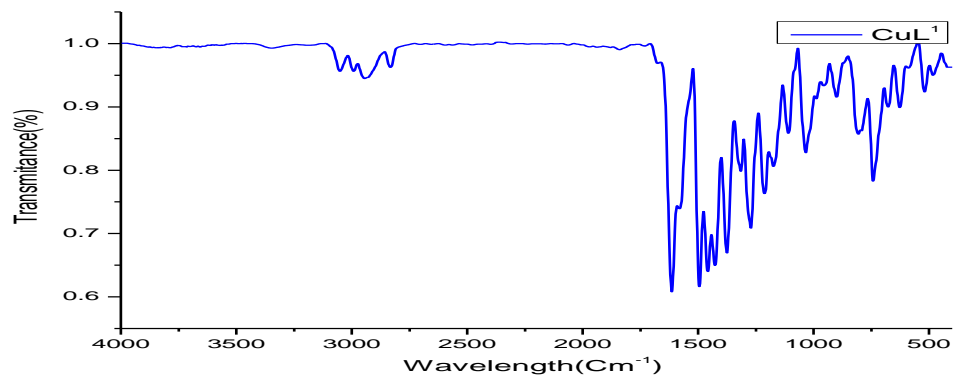
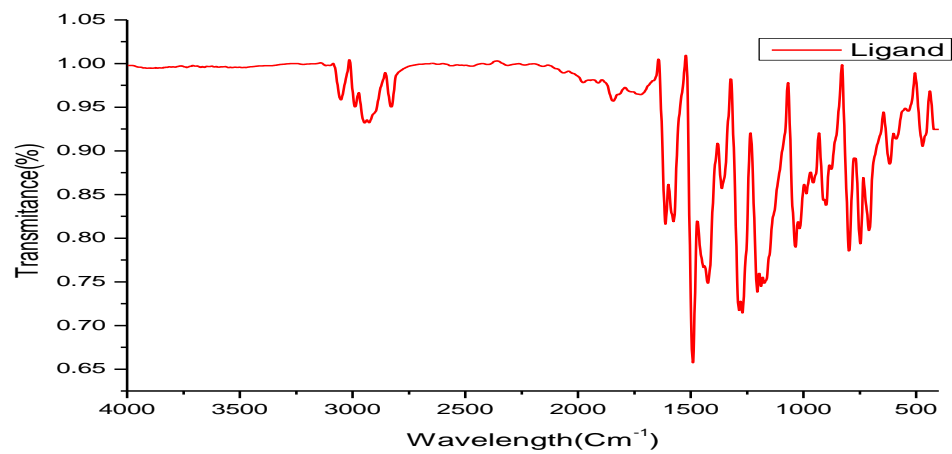
Supplementary Table 8. Anti-diabetic activity data of Schiff base ligand, CuL¹ L² and CuL¹ complexes

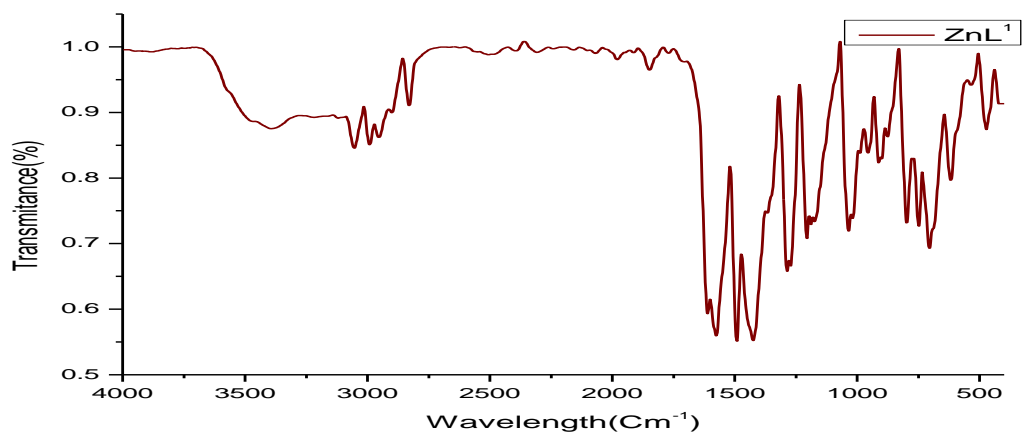
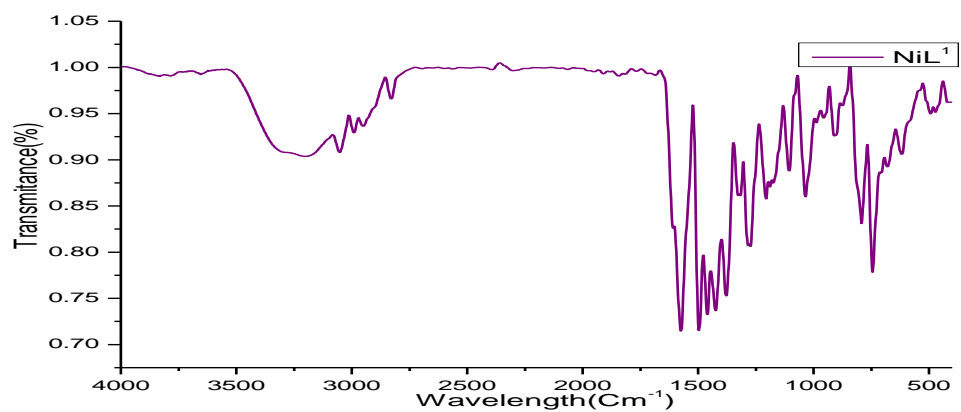
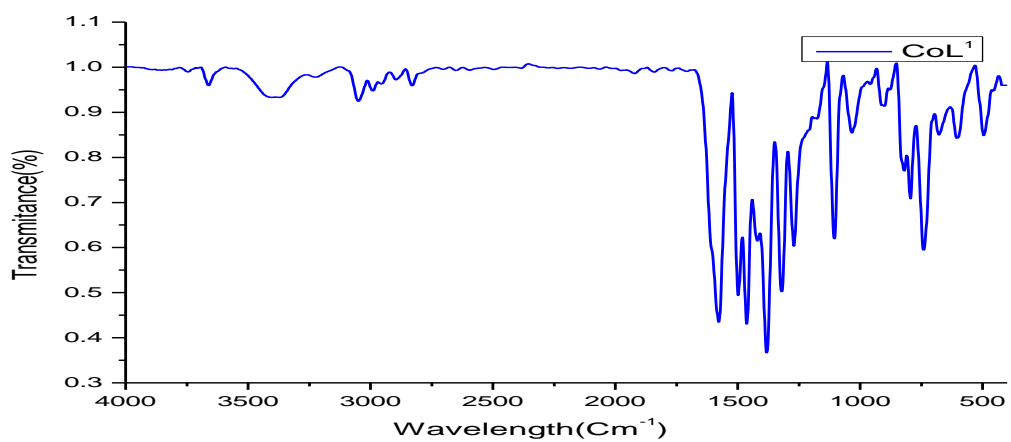
Compounds	Concentration (µg/ml)					IC ₅₀ (µg/ml)
	10	20	30	40	50	
% of inhibition at 540 nm						
Ligand	37.33	42.80	43.16	45.54	48.98	54.80
CuL ¹ L ²	47.08	65.51	68.25	71.34	72.88	3.86
CuL ¹	53.62	61.59	62.66	63.85	66.34	11.91

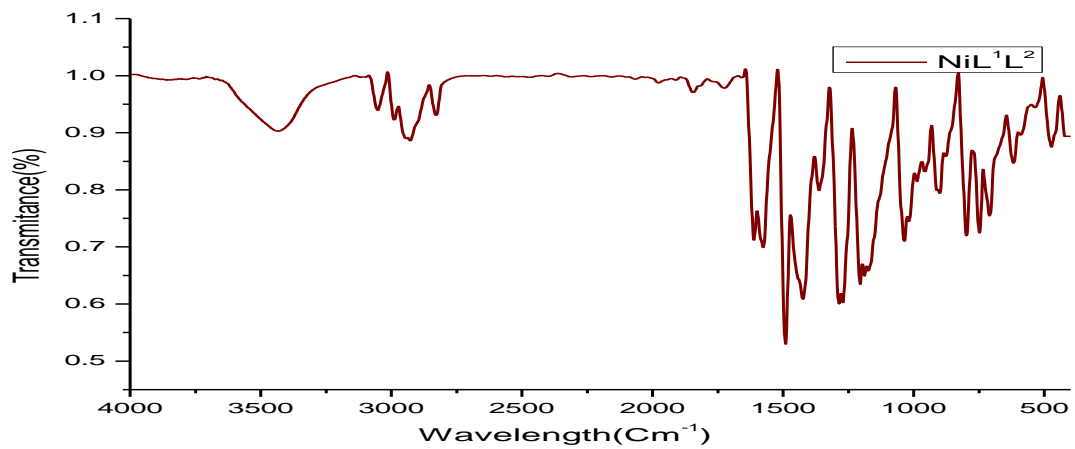
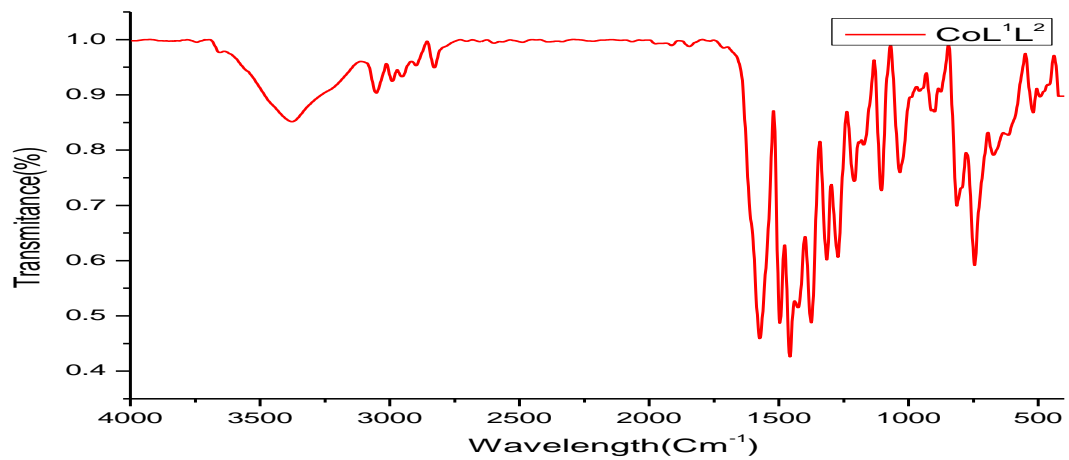
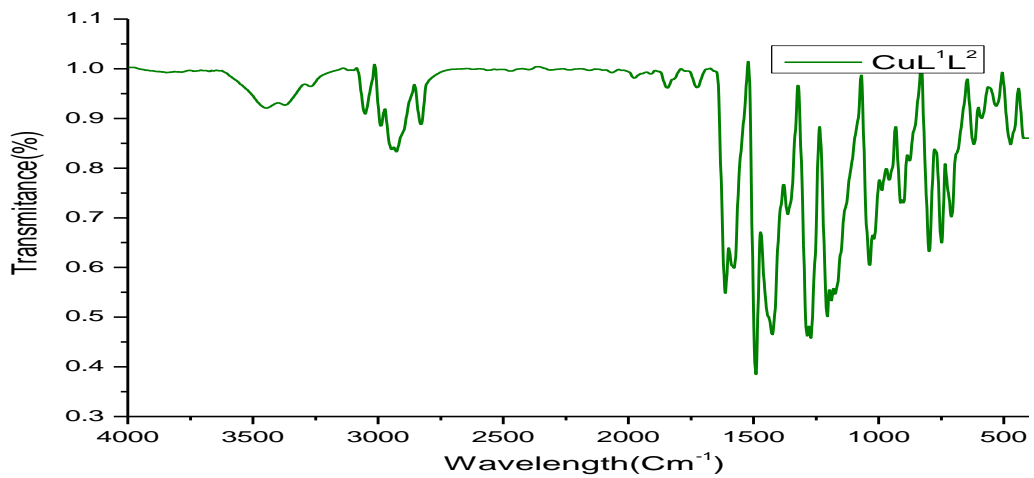
Supplementary Table 9. Anti-Cancer activity data of the Schiff base ligand, CuL¹ L² and CuL¹ complexes

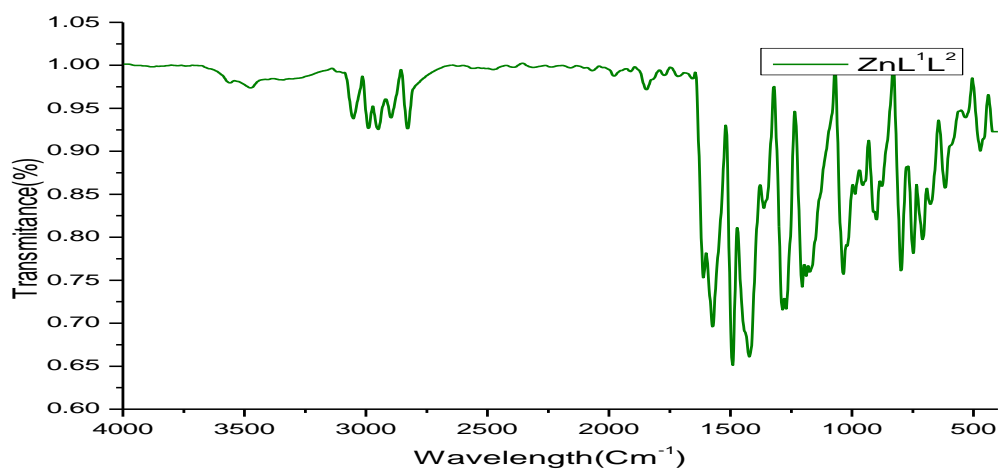
Ligand Concentration (µg / ml)	Ligand Cell Viability %	IC ₅₀ (µg / ml)	CuL ¹ L ² Cell Viability %	IC ₅₀ (µg / ml)	CuL ¹ Cell Viability %	IC ₅₀ (µg / ml)
3.125	97.8545888		89.91646778		97.09697344	
6.25	90.40524434		74.46300716		87.3996294	
12.5	71.81168057	32.63	59.78520286	24.92	70.84620136	30.55
25	46.72228844		37.58949881		48.98085238	
50	33.49225268		24.34367542		26.93020383	

Supplementary Figure 1. FT-IR spectra of L, ML¹ & ML¹L²

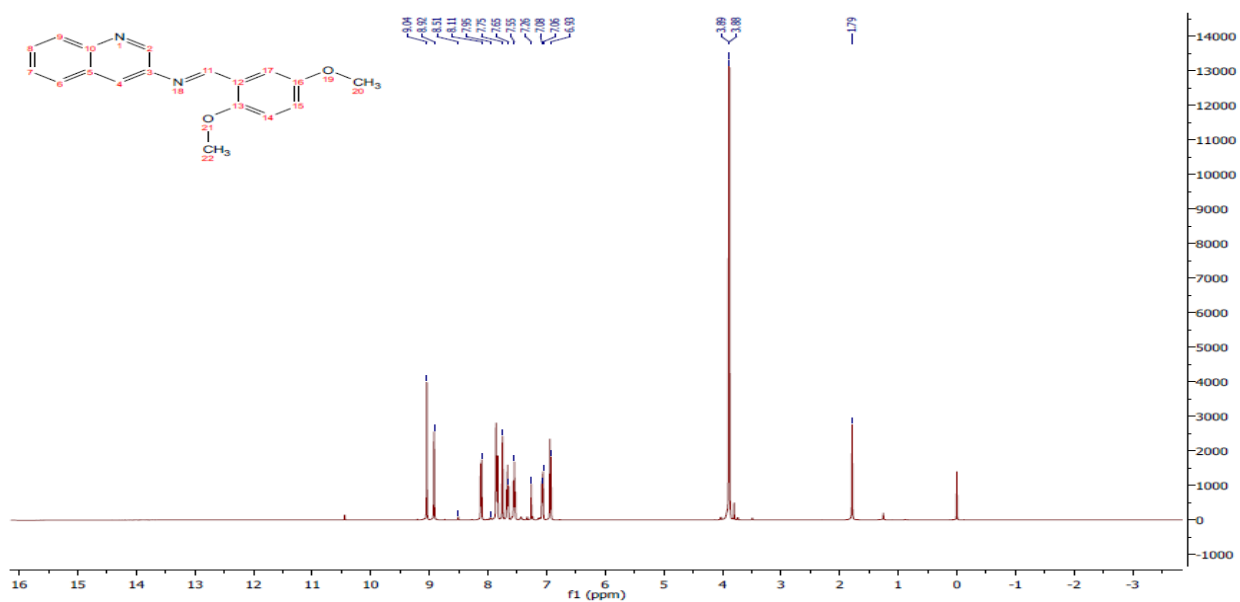




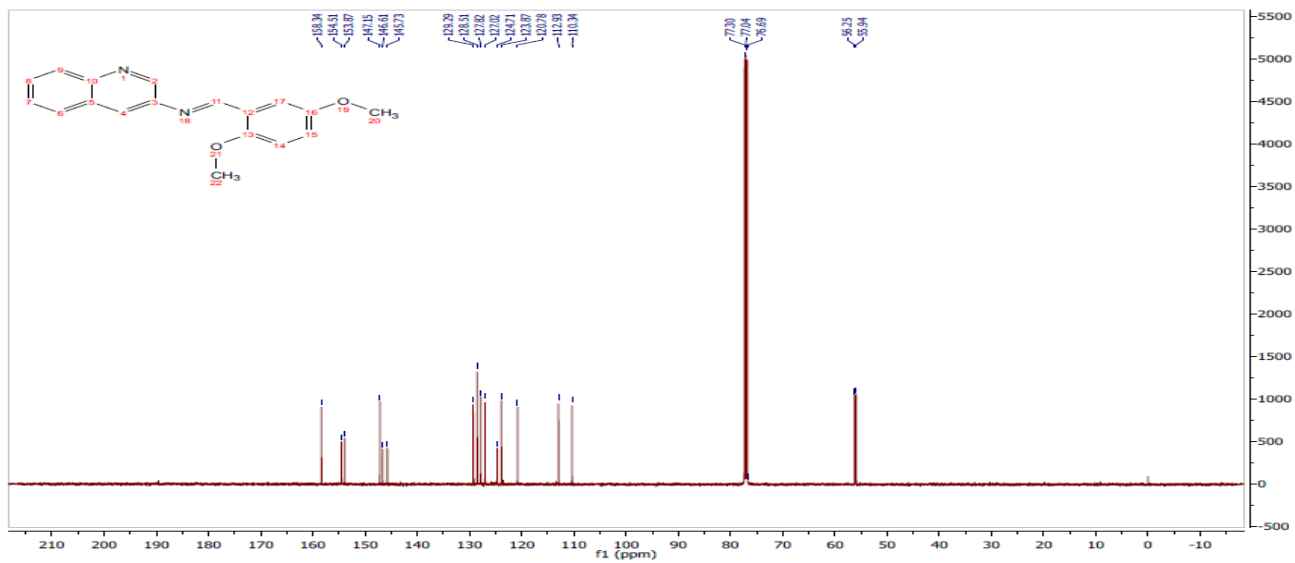




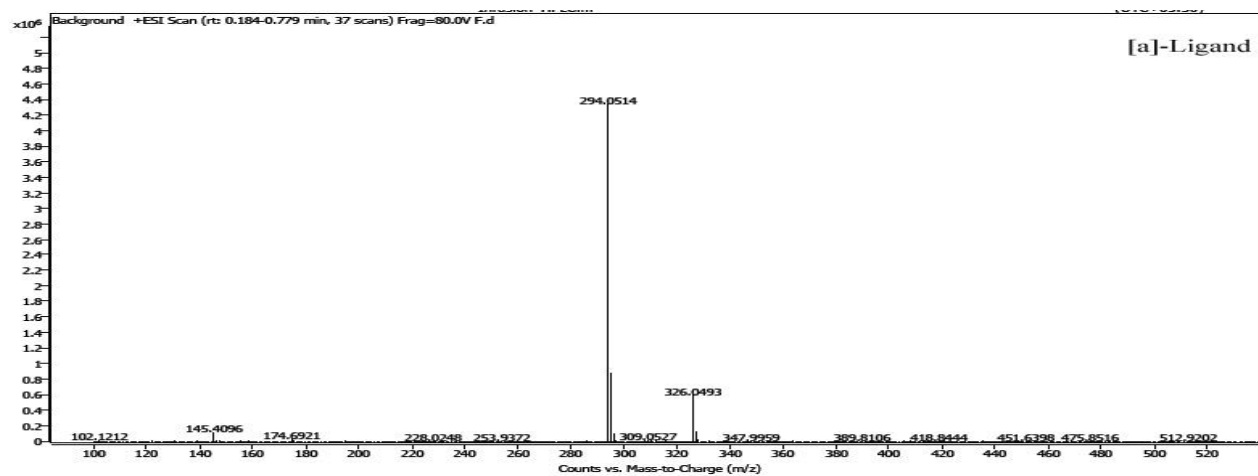
Supplementary Figure 2. 1H NMR of Schiff base ligand (L)

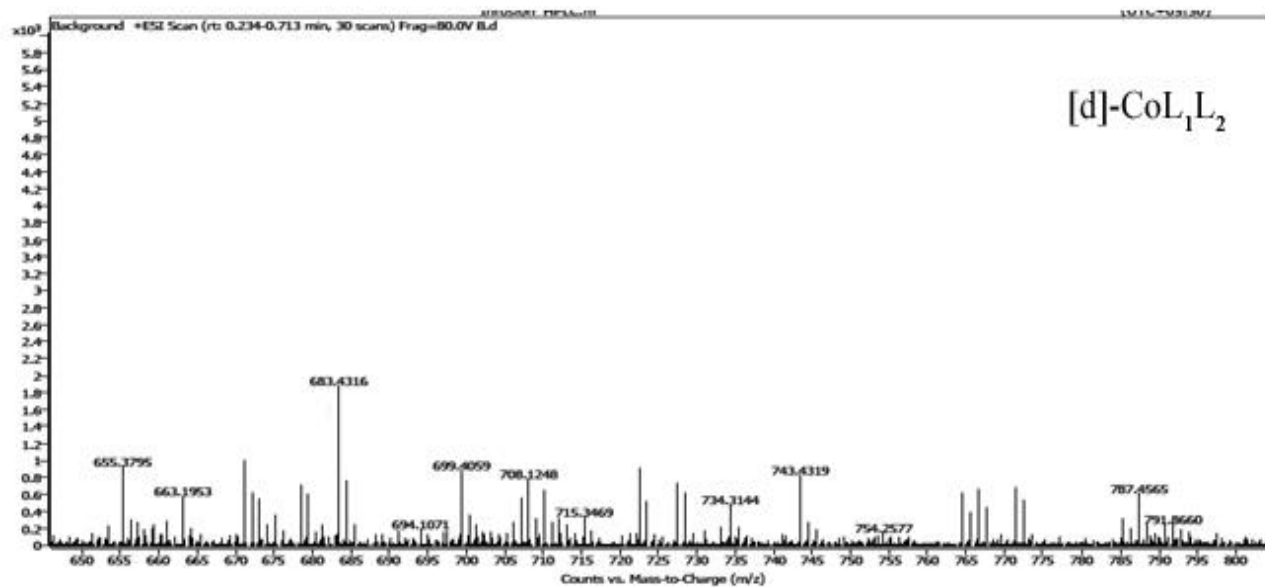
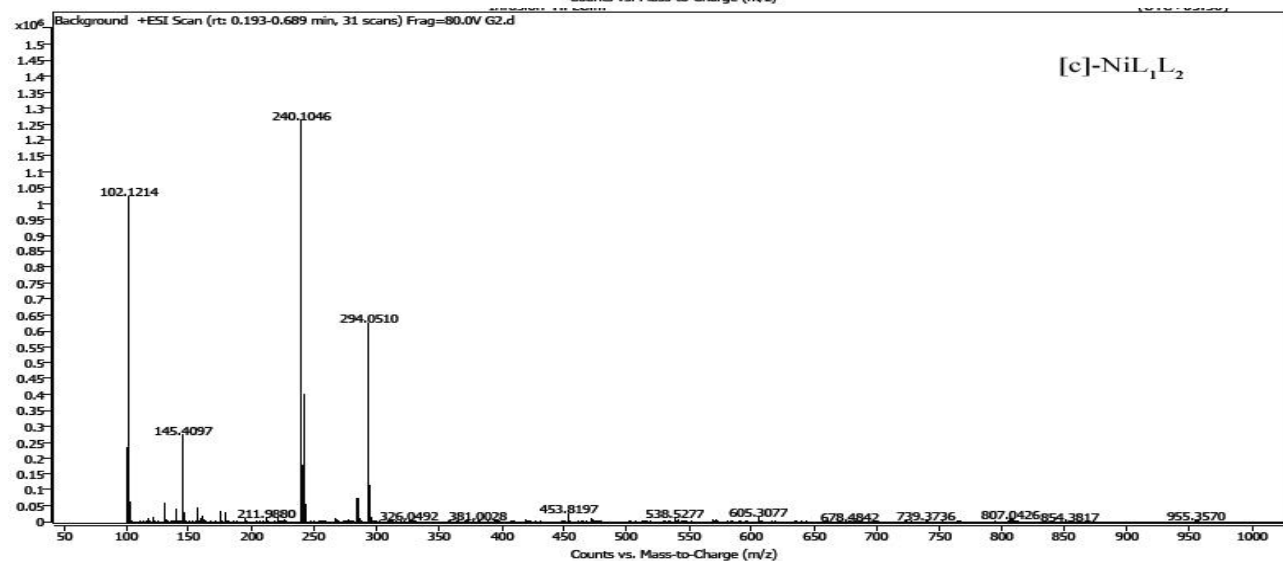
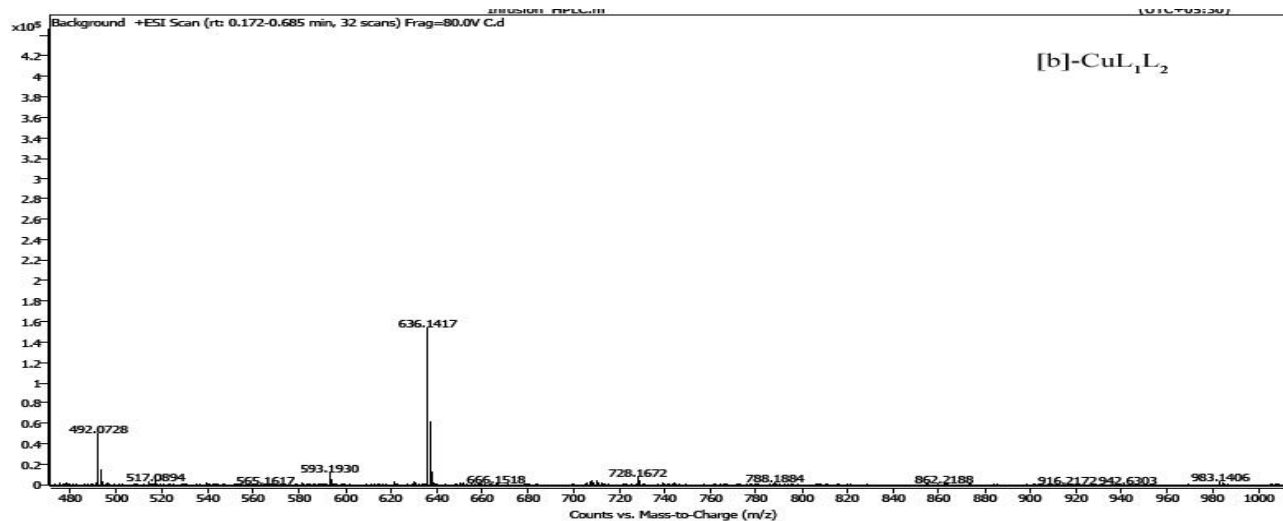


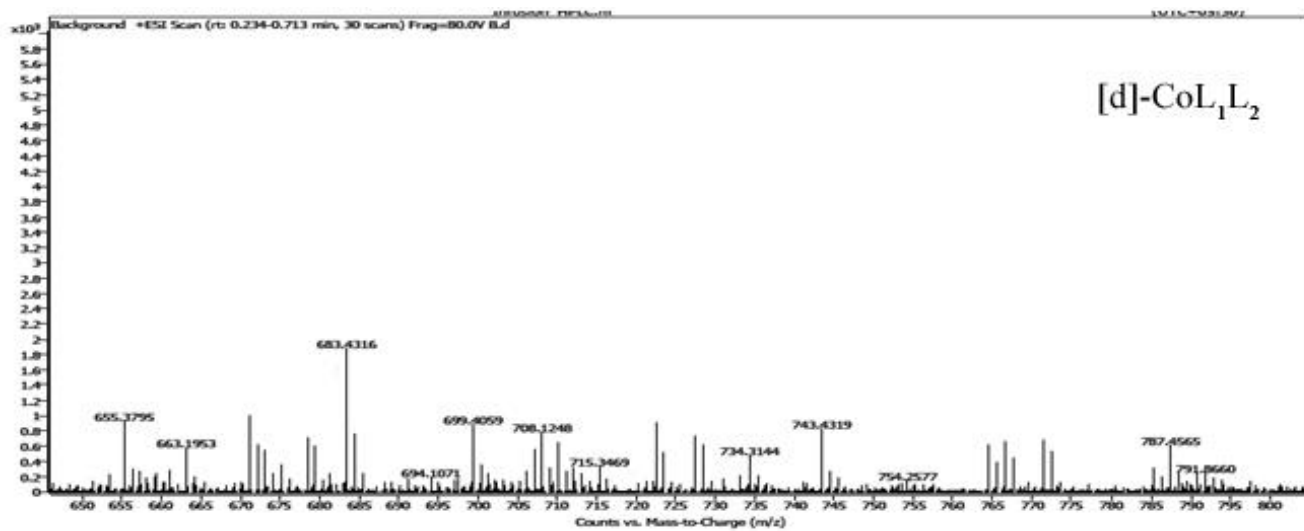
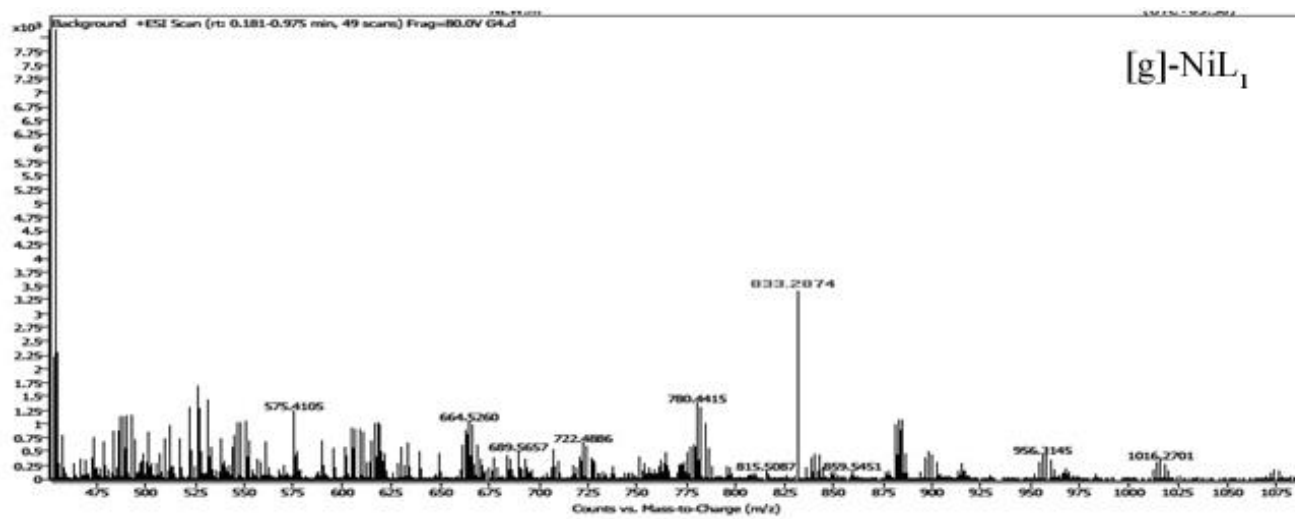
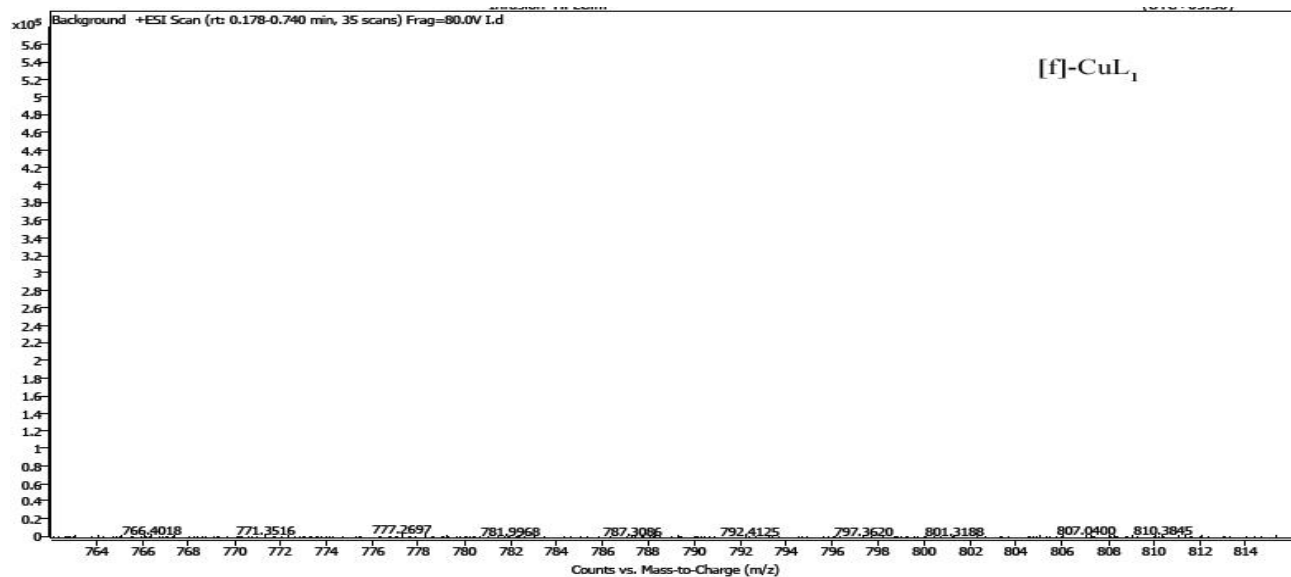
Supplementary Figure 3. ^{13}C NMR for Schiff base ligand(L)

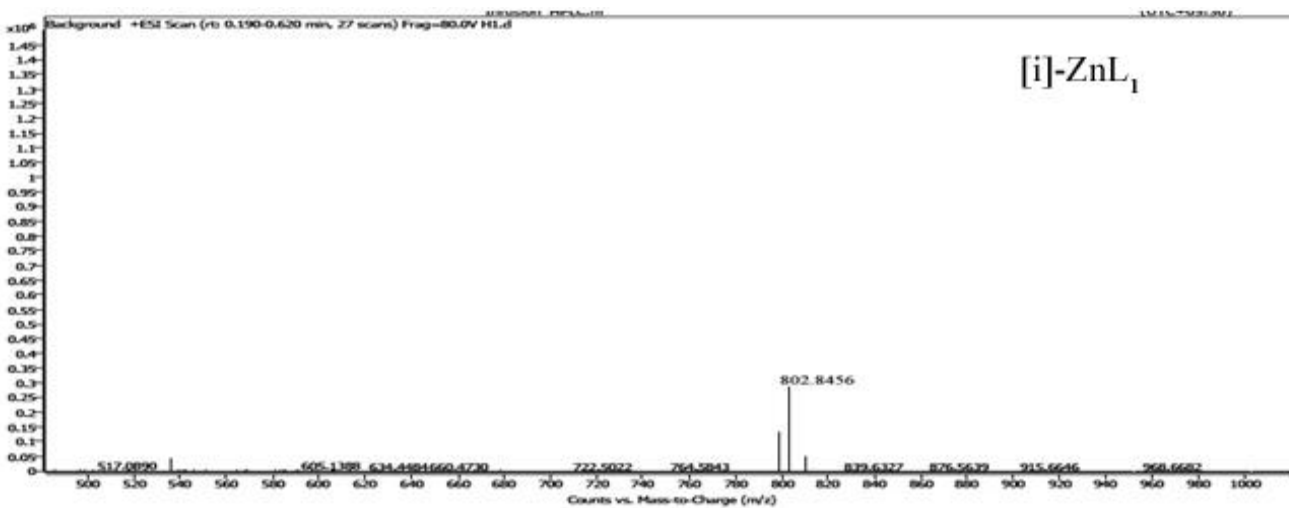


Supplementary Figure 4. Mass spectra of L, ML¹ & ML¹L²

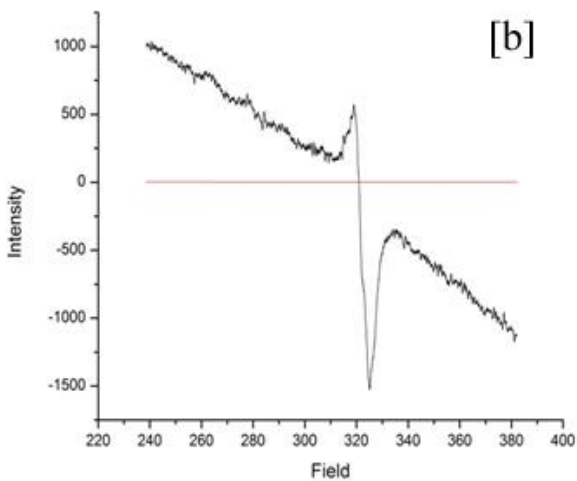
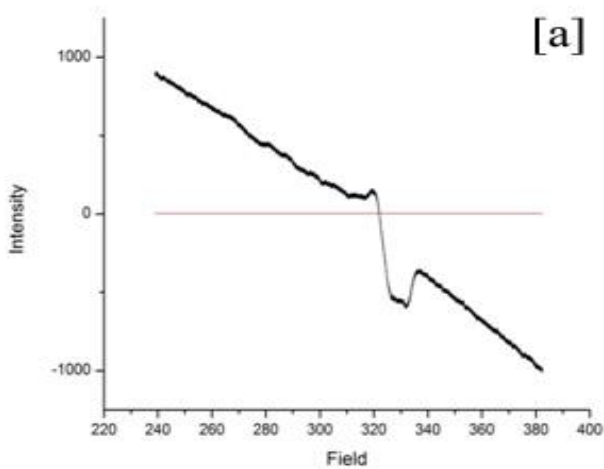








Supplementary Figure 5. EPR spectra for CuL¹ and Cu L¹L² complexes



Supplementary Figure 6. XRD analysis of L, ML¹ & ML¹L²

